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Correlation between the microstructure and electroluminescence properties of Er-doped metal-oxide semiconductor structures

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Optical response of a rare earth (RE)-doped SiO₂ layer is known to deteriorate markedly at room temperature due to RE clustering. The key challenge is therefore to probe the ongoing processes at the microscopic level and the subsequent impact on the luminescence properties with increasing RE concentration. Here, we report how the Er electroluminescence in a metal-oxide-semiconductor structure has been affected by increasing Er content. Our results indicate that the Er oxide clustering is anticipated by the formation of Si-based oxygen-deficiency centers during postimplantation annealing and leads to a strong quenching of the short-wavelength (350–500 nm) Er electroluminescence. © 2009 American Institute of Physics. [DOI: 10.1063/1.3098474]

Although silicon based light emitters have been realized by utilizing nanoscale structures,^{1,2} the efficiency of such devices is far from the practical application. In fact, electrically pumped light-emitting devices (LEDs) are highly desirable from the optoelectronic standpoint. In spite of considerable progress in nanotechnology the introduction of the rare earth (RE) doped metal-oxide-semiconductor LEDs (MOS-LEDs) (Ref. 2) gives an auxiliary platform, where the RE-based electroluminescence (EL) is controlled by the intrashell transitions in partially filled shells.³ For instance, the intra- $4f$ shell transition ($^4I_{13/2} \rightarrow ^4I_{15/2}$) of Er³⁺ can emit an infrared (IR) light at $\sim 1.53 \mu\text{m}$.⁴ We must note here that while most of the intraband transitions are optically forbidden for RE ions, no such constraint exists during electrical pumping; hence, several EL signals can be expected during electron injection. The temperature and concentration quenching of RE luminescence remain the major limiting factors to achieve maximum efficiency at room temperature (RT).² Owing to low solubility of the RE elements in SiO₂,⁵ a significant fraction of RE ions generally prefer to form clusters during postimplantation annealing.⁴ In this letter we demonstrate the evolution of microstructure in Er-doped MOSLEDs as a function of Er concentration, [Er], and underscore its influence on Er electroluminescence.

Among various approaches, ion implantation is recognized as a powerful method owing to its excellent control over distribution of dopants and residual defect.⁴ Here, the Er-implanted MOS structures were fabricated by standard local oxidation of silicon (LOCOS) technology with a 200 nm thick thermally grown SiO₂ layer on n -type Si(100) wafers, where the 250 keV Er ions with fluences in the range of $(1-5) \times 10^{15}$ ions/cm² were implanted using a typical target current density of ~ 200 nA/cm². Initially, the Er concentration was determined to be in between 0.3% and 1.4% at R_p (~ 115 nm) by the SRIM-2006 calculations,⁶ which was later confirmed by Rutherford backscattering spectroscopy measurements. The samples were subjected to furnace annealing for 30 min at 900 °C in nitrogen ambience. ~ 100 nm thick silicon-oxynitride (SiON) layer was deposited on top of the LOCOS structure. In addition, a semitransparent indium-tin oxide and aluminum contacts were sputter

deposited in the front and rear surfaces, respectively. Arrays of circular electrodes (diameter of $\sim 300 \mu\text{m}$) were developed on the front surface by optical lithography. All the RT EL spectra, recorded in constant current mode and under forward bias condition, were detected by a monochromator in combination with a photomultiplier or a liquid nitrogen cooled InGaAs detector. The photomultiplier in combination with a photon counting system were employed to study the time-resolved EL dynamics. Cross-sectional transmission-electron-microscopy (TEM) images were taken by a FEI Titan S/TEM, operating at 300 keV.

Figure 1(a) shows the EL spectra, recorded with a constant current density (J) of ~ 7.1 mA/cm² for the MOSLEDs containing 0.3%, 0.8%, and 1.4% Er. The inset projects another visible range spectrum from a *different* set of MOSLEDs containing 0.5% Er for $J=70.7$ mA/cm². Although the implantations have been carried out at RT, the difference in EL spectra implies that the signal is very sensitive to the sample preparation conditions (such as SiO₂ and SiON layer thicknesses, implantation energy, target current density, Er concentration, etc.). The EL bands peaking at ~ 387 , 412, 437, 478, 527, 551, and 660 nm (marked by vertical lines) can be assigned as the radiative transitions from the $^2G_{11/2}$, $^2H_{9/2}$, ($^4F_{3/2} + ^4F_{5/2}$), $^4F_{7/2}$, $^2H_{11/2}$, $^4S_{9/2}$ and $^4F_{9/2}$ states to the ground $^4I_{15/2}$ state of Er³⁺, respectively.³ Note that the cut-off wavelength of the photomultiplier is ~ 370 nm, which may affect the 387 nm Er peak. The IR peak at ~ 1535 nm is the characteristic feature of the $^4I_{13/2} \rightarrow ^4I_{15/2}$ transition in Er³⁺.⁷ Nevertheless, the 437 and 478 nm EL signals are indistinguishable in the present set [main part of Fig. 1(a)] up to 0.8% Er. Apparently, a broad EL band at ~ 460 nm (indicated by *downward arrow*) that evolves by suppressing the Er EL signals in the range of 350–500 nm for 1.4% Er can be attributed to the Si-based oxygen-deficiency centers (SiODCs).¹ Since the 1535 nm Er EL intensity is almost saturated at $\sim 0.8\%$ Er, MOSLEDs made of 0.8% Er are considered to be the best-emitting electrical structures.

The visible range EL spectrum has been monitored as a function of J for the 0.8% Er-doped MOSLEDs [Fig. 1(b)]. As evident, the intensity of the short wavelength Er peaks, especially the 412 nm EL becomes more prominent with increasing J , signifying efficient excitation of the $^2H_{9/2}$ or the

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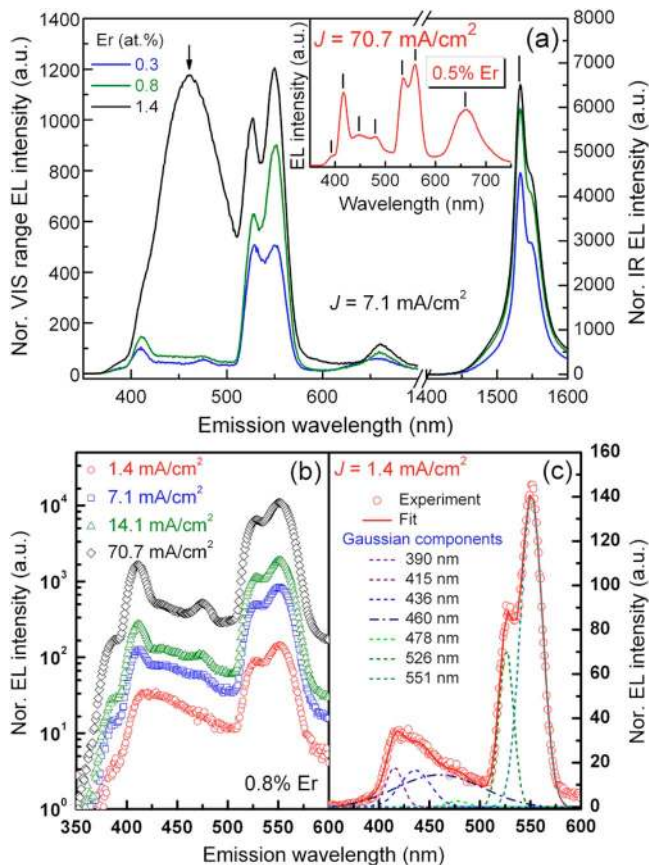


FIG. 1. (Color online) The visible and IR EL spectra recorded from MOSLEDs with 0.3%, 0.8%, and 1.4% Er (a). Another spectrum for 0.5% Er is shown in inset (a) Visible EL spectra for 0.8% Er with increasing J are displayed by open symbols (b). Deconvoluted Er peaks (dashed curves) and the 460 nm band (dash-dot curve) together with fit (solid curve) are superimposed on the experimental data for $J=1.4$ mA/cm² (c).

states above Er³⁺ by hot electrons and subsequent relaxation back to the ground ⁴I_{15/2} state.³ The experimental data for $J=1.4$ mA/cm² is well reproduced using seven Gaussian bands with peak maxima at 390, 415, 436, 460, 478, 526, and 551 nm, respectively [see Fig. 1(c)]. Clearly, except the 460 nm Gaussian band, other components are near to the expected peak positions of Er³⁺ ions, as stated above.³ The fitting procedure, therefore, confirms the coexistence of both Er-related emissions and the SiODC-induced EL at ~460 nm,¹ where the peak intensity for the latter one domi-

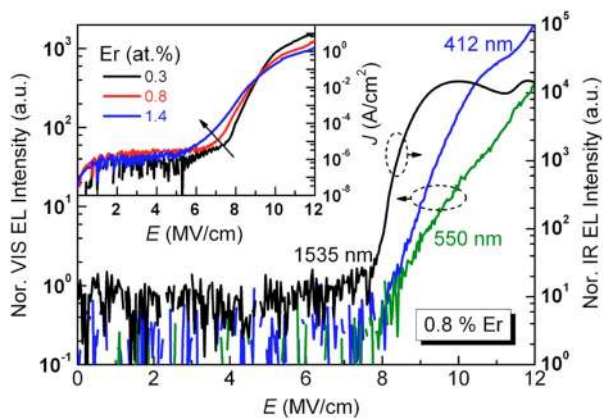


FIG. 2. (Color online) The EL intensity of the 412, 550, and 1535 nm Er peaks with increasing electric field for 0.8% Er. The inset exhibits Er concentration dependent J - E characteristics.

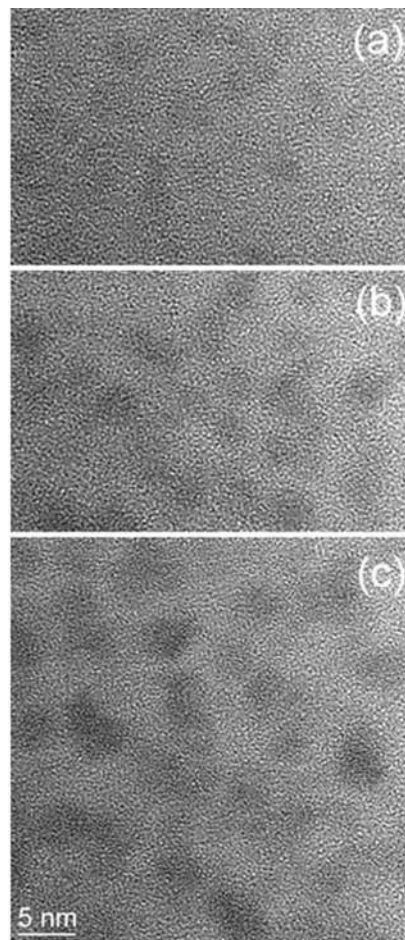


FIG. 3. High-resolution TEM images for 0.3% (a), 0.8% (b) and 1.4% (c) Er. The dark contrast in gray background represents Er oxide clusters. The average cluster size is estimated to be ~2.1 nm (a), 3.8 (b), and 5.3 nm (c) near R_p . All TEM images are in the same scale.

nates further by introducing additional Er [see Fig. 1(a)]. Based on the EL results, we believe that the ion beam induced defects in SiO₂, especially those based on a local deficiency of oxygen, cannot be removed completely during postimplantation annealing.

In order to follow the electric field (E) dependent excitation mechanism of Er³⁺, the 412, 550, and 1535 nm EL bands have been chosen due to their prominent feature. The respective EL intensity dependence is monitored for the MOSLEDs containing 0.3% and 0.8% Er, where the profiles for the latter case are exhibited in Fig. 2. Apparently, the Er³⁺ ions do not emit light below 7.5 MV/cm, while the trends of the ²H_{9/2}→⁴I_{15/2} (blue), ⁴S_{9/2}→⁴I_{15/2} (green), and ⁴I_{13/2}→⁴I_{15/2} (black) EL intensity curves are almost independent to each other. From the energy level standpoint in Er³⁺, it is expected that the first excited state (⁴I_{13/2}) can be excited efficiently by relatively lower E value than the other high energy states. In fact, the 1535 nm EL intensity starts increasing much earlier than the 412 and 550 nm bands and saturates faster. The inset of Fig. 2 summarizes the J - E characteristics of the working devices, confirming the device integrity even after device processing. A small displacement current is measured in lower electric fields, where the experimental data can be fitted by a power law $J=A_1E^m$ for $0.5 < m < 0.7$. This observation is analogous to the Si-rich nitride/Si superstructures,⁸ and hence the electron injection can be explained in the light of space charge limited conduc-

TABLE I. Results of the decay time (τ_d), dispersion factor (β), and the $\sigma\tau$ analysis for different Er EL bands with increasing [Er].

Emission wavelengths (nm)	$\tau_d(\mu\text{s}) \pm 5\%$			$\beta \pm 5\%$			$\sigma\tau(\text{cm}^2 \text{s}) \pm 5\%$		
	Er conc. (at. %)			Er conc. (at. %)			Er conc. (at. %)		
	0.3	0.8	1.4	0.3	0.8	1.4	0.3	0.8	1.4
412	207	267	...	0.56	0.41	...	1.0×10^{-20}	8.2×10^{-21}	...
460	4867	0.81	3.3×10^{-18}
550	232	239	274	0.47	0.44	0.50	2.1×10^{-20}	3.2×10^{-22}	8.8×10^{-21}

tion to the SiON layer in the low electric field region. Here, m depends on the energy width of the trap distribution in SiON, while A_1 is a constant. Above this electric field region, the J - E characteristic shows an almost exponential behavior. As evident, the threshold E moves toward the lower value with increasing [Er], suggesting a decrease in barrier height at the Si/SiO₂ interface.⁹ While the charge transport from the conduction band (CB) of Si to the CB of SiO₂ layer is found to be mainly governed by the Fowler–Nordheim tunneling in higher E values, the contribution from other mechanisms such as Poole–Frenkel emission, hopping, charging, etc., has also been noticed.¹⁰

It is known that the SiO₂ network is destroyed during ion implantation and displaces Si and oxygen (O) atoms. The displaced Si and O atoms may either be absorbed into the local SiO₂ network or diffused to the impurity (Er) sites for achieving a stable composition. According to the formation enthalpies of SiO₂ (−856.3 kJ/mol) and Er₂O₃ (−1808.7 kJ/mol),¹¹ a close competition in reconstructing SiO₂ and Er₂O₃ configurations is expected. Indeed, a considerable fraction of Er atoms are involved in developing amorphous Er oxide clusters^{4,5} (Fig. 3) during annealing and eventually trigger the formation of SiODCs. However, due to the presence of the surrounding SiO₂ networks, it is impossible to draw a conclusion over the exact composition of the Er oxide clusters. Discernibly, the average size of such Er oxide clusters gradually increases from ~ 2.1 to 5.3 nm at R_p with increasing [Er]. Note that the Er oxide clusters are more or less distributed over a region of ~ 80 nm in the central part of the SiO₂ layer. The decay time (τ_d) of the prominent EL peaks (Table I) has also been estimated using a stretched exponential¹² $I(t) \propto \exp[-(t/\tau)^\beta]$, where β is a dispersion factor indicating a spread of Er lifetime (τ) due to the presence of different Er sites.¹³ The product of the excitation cross-section (σ) and τ has been calculated according to Ref. 14, evidencing overall quenching of σ with increasing cluster size. Interestingly, the decay times for the blue and green EL bands are found to increase with increasing [Er], which is just opposite to the known phenomenon of concentration quenching.¹⁵ Since the contribution of the SiODCs dominates over Er emission [see Fig. 1(a)] and the average size of Er oxide clusters is found enhancing simultaneously with increasing [Er] (Fig. 3), it seems that a delicate balance between the SiODCs near the cluster surface and the cluster size itself plays the pivotal role in controlling EL. Indeed, the electric field would be lower within an Er oxide cluster than in SiO₂ as the dielectric constant of the Er₂O₃ (~ 13) (Ref. 16) is higher than that of SiO₂ (3.9).¹⁰ The relatively weak field in an Er oxide cluster cannot accelerate electrons efficiently in order to excite the Er³⁺ ions. Hence, the Er³⁺ ions residing near to the core region of a cluster are assumed to be

inactive, whereas the ions situated at the surface can predominantly take part in EL. Although the increase in cluster size reduces the number of surface Er³⁺, the subsequent increase in cluster density by enhancing [Er] (Fig. 3) is expected to compensate the lack of excitable Er³⁺ ions. Moreover, Er³⁺ ions in the core region of a cluster might be excited electrically if relatively high electric field, like above 11 MV/cm, is applied for the blue and IR EL profiles (Fig. 2).

In conclusion, we have demonstrated the EL properties of the Er-doped MOSLEDs with increasing [Er] up to 1.4% by lowering the barrier height at the Si/SiO₂ interface. The variation in EL output has been illustrated on the ground of increasing average sizes of the Er oxide clusters as a function of [Er], leading to the formation of the SiODCs in SiO₂. In fact, the SiODC induced 460 nm EL dominates over the short wavelength (350–500 nm) Er signals with increasing [Er].

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